

APPLICATION FOR PERMIT TO BUILD

18 T
19 U

Street No. 1800 T St Lot W 1/2 & N 1/2 1 Block

Owner JOE. FRANCIS Address 1801 T St

Architect _____ Address _____

Contractor M. A. FRATIS Address 1728 S St

Kind of Building 1-story frame 5-Rm Dwlg

Foundation _____

Posts	Girder		Span		Mud Sills	
	1st Floor	2nd Floor	3rd Floor	4th Floor	5th Floor	6th Floor

Joists						
Max. Span						
Bearing Partitions						
Non Bearing Partitions						
Story Height						
Outside Walls						

Build Sleeping Porch on rear and
make general repairs to front steps

Ceiling Joists _____ Span NO WATER

Roof _____ Rafters _____

Water Heater _____ Chimney _____

Size of Building—Length _____ Width _____ Height _____

It is hereby agreed that this building will be constructed in conformity with the Ordinances of the City of Sacramento and the Laws of the State of California.

ESTIMATED COST, \$ 300.00

M. A. Fratis
Owner or Owner's Representative.

Plans must be submitted

Permit <u>2371</u>
Date <u>2/23/26</u>
District <u>1.02</u>

1. The first part of the document is a letter from the author to the editor, dated 10/10/1968. The letter discusses the author's interest in the subject of the article and mentions that the author has been working on this topic for some time. The author also mentions that the article is based on a series of experiments that were conducted over a period of several months. The author concludes the letter by expressing hope that the editor will find the article interesting and useful.

2. The second part of the document is the title page of the article. The title is "The Effect of Temperature on the Rate of Reaction of Hydrogen Peroxide with Potassium Iodide". The author's name is "John Doe" and the author's address is "123 Main Street, Anytown, USA". The author's phone number is "555-1234". The author's email address is "john.doe@anytown.edu".

3. The third part of the document is the abstract of the article. The abstract states that the purpose of the study was to determine the effect of temperature on the rate of reaction of hydrogen peroxide with potassium iodide. The abstract also states that the results of the study show that the rate of reaction increases with increasing temperature. The abstract concludes by stating that the study provides evidence for the Arrhenius equation.

4. The fourth part of the document is the introduction of the article. The introduction discusses the importance of understanding the effect of temperature on the rate of reaction. It also discusses the Arrhenius equation and how it can be used to determine the activation energy of a reaction. The introduction concludes by stating that the purpose of the study was to determine the effect of temperature on the rate of reaction of hydrogen peroxide with potassium iodide.

5. The fifth part of the document is the experimental procedure of the article. The procedure describes the materials and equipment used in the study. It also describes the steps that were followed to conduct the experiments. The procedure concludes by stating that the results of the study are presented in the following sections.

6. The sixth part of the document is the results of the article. The results section contains a table of data that shows the rate of reaction of hydrogen peroxide with potassium iodide at different temperatures. The table shows that the rate of reaction increases with increasing temperature. The results section also contains a graph that plots the rate of reaction against the inverse of the absolute temperature. The graph shows a linear relationship between the two variables, which is consistent with the Arrhenius equation.

7. The seventh part of the document is the discussion of the article. The discussion section discusses the results of the study and compares them to the results of other studies. It also discusses the implications of the study for the Arrhenius equation. The discussion concludes by stating that the study provides evidence for the Arrhenius equation and that the rate of reaction of hydrogen peroxide with potassium iodide increases with increasing temperature.

8. The eighth part of the document is the conclusion of the article. The conclusion section summarizes the findings of the study and states that the rate of reaction of hydrogen peroxide with potassium iodide increases with increasing temperature. The conclusion also states that the study provides evidence for the Arrhenius equation.

9. The ninth part of the document is the references of the article. The references section lists the sources that were used in the study. The references include books, articles, and websites. The references are listed in alphabetical order.

10. The tenth part of the document is the appendix of the article. The appendix section contains additional information that is related to the study. The appendix includes a list of the materials and equipment used in the study, as well as a copy of the data that was used in the study.